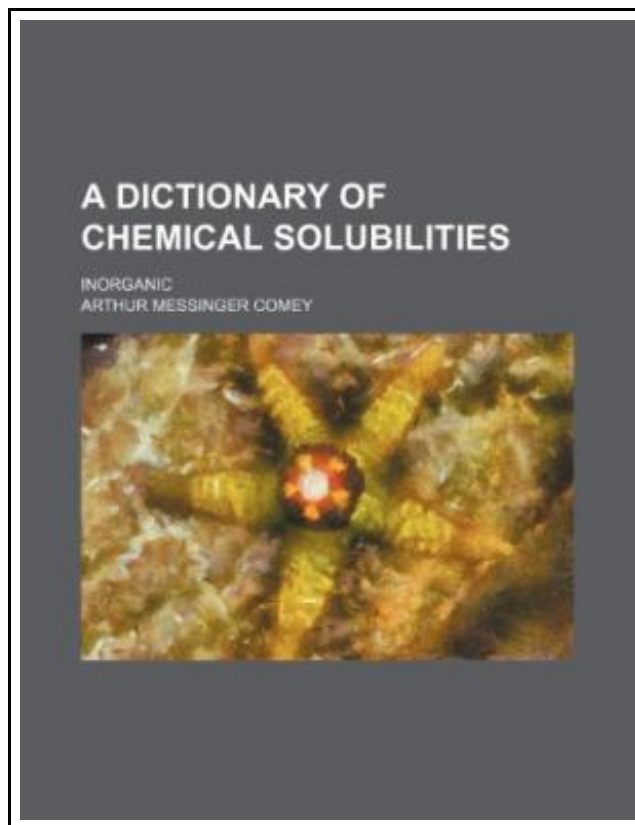


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A DICTIONARY OF CHEMICAL SOLUBILITIES; INORGANIC



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1896 Excerpt: . we have an intermittent passivity. If a part of a piece of iron is immersed in strong acid, the whole of it is made passive. This is explained by the NO spreading over the whole surface by capillarity. The passivity ceases when the Fe is placed in dil. acid, after a longer or shorter time, according to the dilution of the acid, --when the acid has sp. gr. = 1-30, after 11 days 1-28 5, 1-26 32 hours 1-16 12 Iron may also be made passive by long standing in NO gas under pressure. (Varenne, C. R. 90. 998.) Fe is made passive by a coating of Fe₃O₄, not by NO. (Schonbein, Pogg. 39. 342.) (Beetz, Pogg. 67. 286.) (Ramann, B. 14. 1430.) Passivity may also be caused by NH₄N₃ + Aq, ammoniacal AgN₃ + Aq, Fe(N₃)₃, Fe(N₃)₂, Al(N₃)₃, Co(N₃)₂, Ni(N₃)₂, etc. + Aq instead of HN₃ + Aq. (Ramann, B. 14. 1933.) Hardly attacked by either dil. or conc. acids when they are under high pressure. (Cailletet, C. R. 68. 395.) Iron is dissolved by HN₃ + Aq, even when very conc., but no gas is evolved and the process is very slow. HN₃ + Aq of the following sp. gr. dissolves the following amts. from strips of pure Fe. (Gautier and Charpy, C. R. 113. 1451.) Not attacked by alkalies. Sol. in NaOH + Aq (34) when air is blown through the liquid. (Zirnite, Ch. Ztg. 12. 355.) NaOH + Aq attacks...



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